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Preparation of superconducting Tl-Ba-Ca-Cu-O thin films by Tl2 O3 vapor processing

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[54] PREPARATION OF SUPERCONDUCTING TL-BA-CA-CU-O THIN FILMS BY TL2O3 VAPOR PROCESSING

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both of Fayetteville, Ark.

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Rock, Ark.

[*] Notice: The portion of the term of this patent

subsequent to Mar. 17, 2009 has been

disclaimed.

[21] Appl. No.: 559,900

[22] Filed: Jul. 30, 1990

Related U.S. Application Data

[63] Continuation of Ser. No. 308,896, Feb. 9, 1989, Pat. No. 4,997,811, which is a continuation-in-part of Ser. No. 236,507, Aug. 25, 1988.

[51] Int. Cl.⁵ B05D 5/12

U.S. Cl. 505/1; 505/742; 427/62; 427/126.3; 427/248.1; 427/343

[58] Field of Search 505/1, 731, 732, 783, 505/742; 427/62, 63, 126.3, 248.1, 255, 333,

343, 53.1

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Primary Examiner-Shrive Beck Assistant Examiner-Roy V. King Attorney, Agent, or Firm-Hermann, Ivester, Hill, Van Santen, Steadman & Simpson

[57] **ABSTRACT**

A process of making high temperature Tl-based superconductors is disclosed. The process includes the steps of reacting solid Ba-Ca-Cu-oxides with Tl₂O₃ vapor. The process allows high quality Tl-based superconductors to be easily fabricated.

11 Claims, 7 Drawing Sheets

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FIG. 1

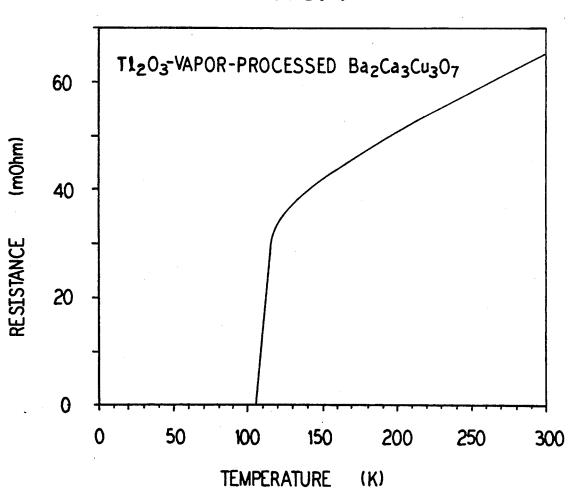


FIG. 2

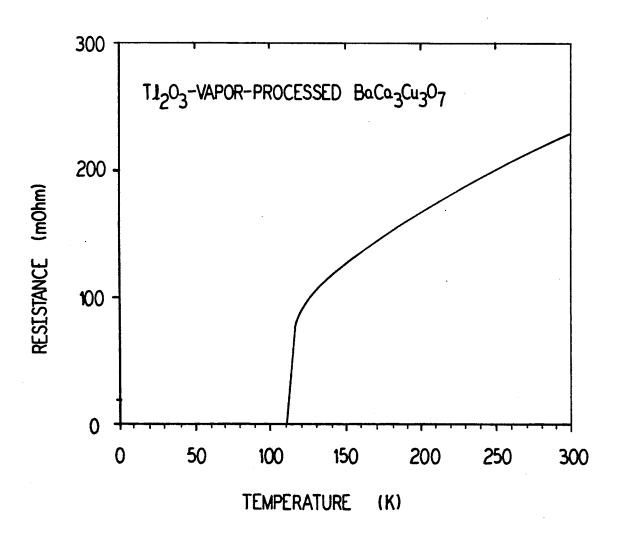


FIG. 3

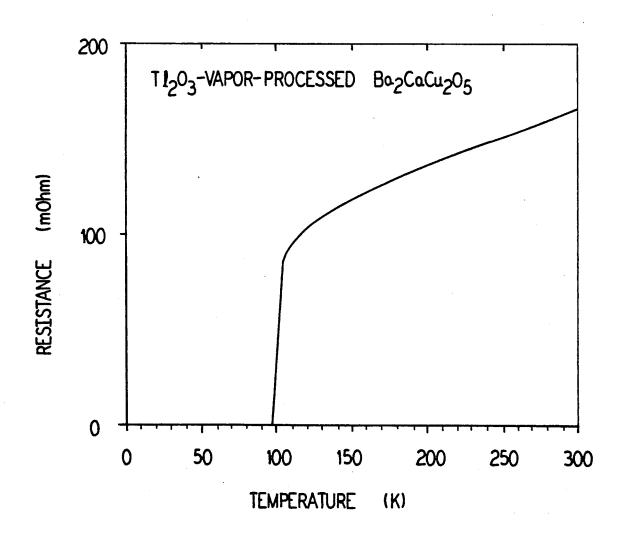


FIG. 4

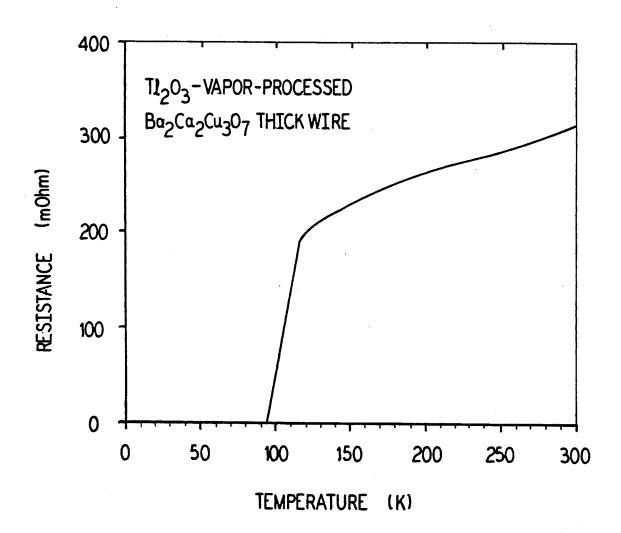


FIG. 5

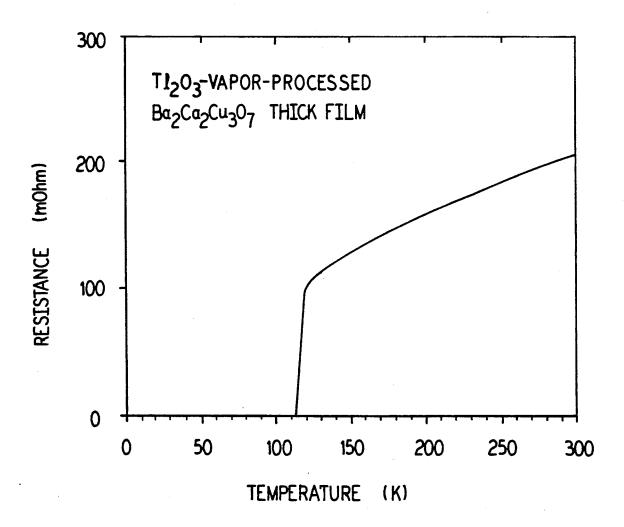
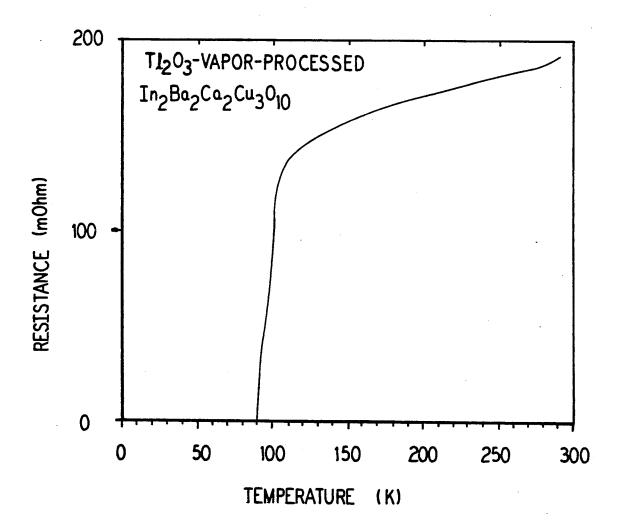
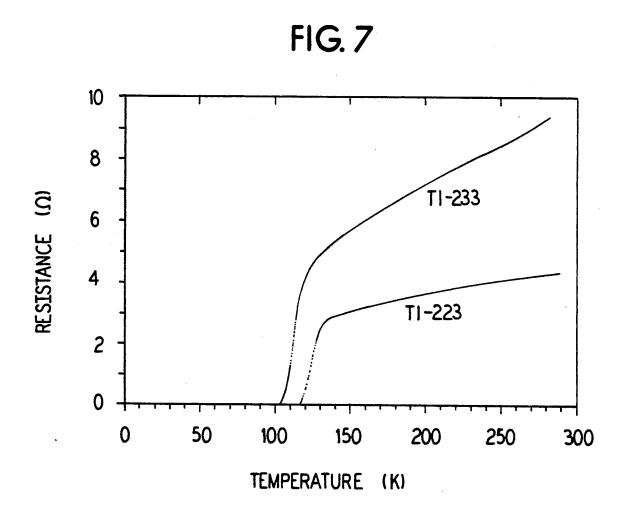


FIG. 6





PREPARATION OF SUPERCONDUCTING TL-BA-CA-CU-O THIN FILMS BY TL₂O₃ VAPOR PROCESSING

This is a continuation of application Ser. No. 308,896 filed Feb. 9, 1989, now U.S. Pat. No. 4,997,811, which is a continuation-in-part of U.S. patent application Ser. No. 236,507 filed on Aug. 25, 1988.

BACKGROUND OF THE INVENTION

The present invention relates generally to high temperature superconductors More specifically, the present invention relates to the fabrication of the high temperature Tl-Ba-Ca-Cu-O superconductors.

U.S. patent application Ser. No. 155,247 now U.S. Pat. No. 4,962,083, filed in the name of the inventors, of the present patent application discloses, in part, Tl-Ba-Ca-Cu-O superconductors that were discovered by the present inventors. These Tl-Ba-Ca-Cu-O superconduc- 20 tors have a transition temperature up to and above 120 K, this transition temperature, the inventors of the present patent application believe, is the highest to date among all existing high temperature superconductors

For useful operation, the transition temperature of a 25 superconductor typically must be at least 1/3 higher than the temperature of operation. For this reason, the Tl-Ba-Ca-Cu-O system has been called the first real liquid nitrogen temperature superconducting system.

The present existing procedures of preparing Tl- 30 based superconductors involve the mixing of all component elements followed by a final sinter. Since Tl₂O₃ evaporates easily, the quality of the Tl-samples is not easily controlled. In addition, in view of its toxicity, Tl₂O₃ presents an additional set of difficulties during the 35 fabrication of these types of superconductors.

An improved method for making Tl-Ba-Ca-Cu-O superconductors would be desirable for many reasons such a process would: (1) allow Tl-based superconductors to be easily constructed ±n the forms of complex 40 bulk components, wires and fibers, and thick and thin films; (2) minimize the toxicity problem caused by Tl compounds; and (3) provide low cost processing and manufacturability.

making Tl-based high temperature superconductors.

SUMMARY OF THE INVENTION

The present invention provides a new process for making Tl-based superconductors The process includes 50 the step of reacting vapor phase Tl₂O₃ and solid Ba-Ca-Cu-oxides. Pursuant to the method of the present invention the Tl-based superconductors are produced in approximately two steps. The first step is the preparation ing of the Ba-Ca-Cu-oxides by the use of Tl2O3 vapor to form high quality Tl-based superconductors. Pursuant to the present invention, the method of making Tl-based superconductors is simplified and only requires the vapor phase treatment.

Accordingly, an advantage of the present invention is that it provides a method which can easily produce Tl-based superconductors.

A further advantage of the present invention is to 65 provide high quality Tl-based superconductors.

A still further advantage of the present invention is that it provides a method which can make Tl-based

superconductors in the forms of bulk materials, wires or fibers, thick and thin films.

Furthermore, an advantage of the present invention is that certain elements can be added into the precursor 5 Ba-Ca-Cu-oxides to satisfy specific needs for various

An additional advantage of the present invention is that the Ba-Ca-Cu-oxides can be made in the recrystalline form of a melt.

Still another advantage of the present invention is that Tl₂O₃-vapor-processing can be carried out in closed containers.

Yet another advantage of the present invention is that it provides a method of making Tl-based superconduc-15 tors which minimizes contamination by Tl compounds.

Additional advantages and features of the present invention are described in and will be apparent from, the detailed description of the presently preferred embodiments and the drawings.

BRIEF DESCRIPTION OF THE DRAWING

FIG. 1 illustrates the relationship of electrical resistance versus temperature for a Tl₂O₃-vapor-processed Ba₂Ca₂Cu₃O₇ sample made pursuant to the method of the present invention.

FIG. 2 illustrates the relationship of electrical resistance versus temperature for a Tl₂O₃-vapor-processed BaCa₃Cu₃O₇ sample made pursuant to the method of the present invention.

FIG. 3 illustrates the relationship of electrical resistance versus temperature for a Tl₂O₃-vapor-processed Ba₂CaCu₂O₅ sample made pursuant to the method of the present invention.

FIG. 4 illustrates resistance-temperature dependence for a Tl2O3-vapor-processed Ba2Ca2Cu3O7 thick wire sample made pursuant to the method of the present invention.

FIG. 5 illustrates resistance-temperature dependence for a Tl₂O₃-vapor-processed Ba₂Ca₂Cu₃O₇ recrystallized thick film sample made pursuant to the method of the present invention.

FIG. 6 illustrates resistance as a function of temperature for a Tl₂O₃-vapor-processed In₂Ba₂Ca₂Cu₃O₁₀ Accordingly, there is a need for a new process of 45 sample made pursuant to the present invention.

FIG. 7 illustrates the relationship of electrical resistance versus temperature for a Tl₂O₃ vapor processed Ba₂Ca₂Cu₃O₇ thin film (Tl-223) made pursuant to the present invention and a Tl₂O₃ vapor processed Ba₂Ca₃. Cu₃O₈ thin film (T1-223) made pursuant to the present

DETAILED DESCRIPTION OF THE PRES-ENTLY PREFERRED EMBODIMENTS

The present invention provides a method for making of Ba-Ca-Cu-oxides; and the second step is the process- 55 high quality Tl-Ba-Ca-Cu-O superconductors. The present invention is based on reactions between vapor phase Tl₂O₃ and solid Ba-Ca-Cu-oxides. The fabrication procedure for the making Tl-Ba-Ca-Cu-O superconductors according to the present invention can be divided manufacture of Ba-Ca-Cu-oxides and a final Tl₂O₃ 60 into two steps: (1) preparation of Ba-Ca-Cu-oxides; and (2) Tl₂O₃-vapor-processing of the Ba-Ca-Cu-oxides to form Tl-Ba-Ca-Cu-O superconductors The present invention simplifies the fabrication of Tl-Ba-Ca-Cu-oxides to the fabrication of Ba-Ca-Cu-oxides, and minimizes problems caused by the toxicity and volatility of Tl compounds. The present invention allows high quality Tl-based superconductors to be easily made in the forms of complex bulk components, wires and fibers, and

thick and thin films, and provides low cost processing and manufacturability of Tl-based superconductors.

Pursuant to the method of the present invention, first Ba-Ca-Cu-oxides are produced. Preferably compounds, the Ba-Ca-Cu-oxides are produced by grinding and 5 mixing one of the following groups of: BaCO₃, CaO, CuO; BaCO₃, CaCO₃, CuO; BaO₂, CaO₂, CuO; or BaO₂, CaO, CuO. In a preferred embodiment, the molar ratio is 2:2:3, respectively, of the components. In an embodiment, the molar ratio of BaCO₃, CaO, CuO is 10 surface, which had an onset temperature of above 120 k 1:3:3

Preferably, after the grinding and mixing the powder is heated. Preferably the powder is heated to approximately 925° C. for approximately 24 to about 48 hours with intermediate grindings.

In an embodiment the resultant powder has the following nominal composition:

BaCaxCuyO1+x+y

wherein:

 $O \le X \le 100$; and $O \le y \le 100$.

In an embodiment, the powder is then pressed into a pellet.

After the powder is prepared, or pressed into a pellet, it can then be reacted with Tl₂O₃ vapor. The Tl₂O₃ vapor can be produced by heating Tl₂O₃ with the Ba-Ca-Cu-oxides or heating any Tl-containing compound that can form Tl₂O₃ vapor upon being heated in oxygen 30 to, for example, approximately 900° C.

In an embodiment, prior to being reacted with Tl₂O₃ vapor, the powder is mixed with a compound chosen from the group consisting of: Tl₂O₃, MgO, AgO, K₂O, Na₂O, ZrO₂, In₂O₃, HgO, Bi₂O₃, PbO₂, and KCl. The 35 resultant mixture is then reacted with Tl₂O₃ vapor.

In an embodiment of the method of the present invention, a Ba-Ca-Cu-oxide film is created and processed with Tl₂O₃ to creating a superconducting thin film.

By way of example and not limitation, examples of 40 the of the present invention process of making high quality Tl-Ca-Ba-Cu-O superconductors will now be given.

EXAMPLE 1

- A. The following reagents were utilized:
 - 1. Tl₂O₃,
 - CaO,
 - 3. BaCO₃,
 - 4 CuO.
- B. The following procedure was followed:
- 1. A mixture of a two molar portion of BaCO₃, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours 55 (with several intermediate grindings) to obtain an uniform black Ba₂Ca₂Cu₃O₇ powder.
- 2. The resulting Ba₂Ca₂Cu₃O₇ powder was completely ground, and pressed into a pellet.
- 3. The pellet was heated at about 925° C. in a tube 60 furnace for about 10 minutes.
- 4. The heated pellet was then taken out of the furnace and cooled in air to room temperature.
- 5. A small amount (approximately 0.1 to about 0.2 gm) of Tl₂O₃ was put in a platinum boat, and the 65 platinum boat was placed in a quartz boat.
- 6. The cooled pellet was placed over the platinum boat.

- 7. The quartz boat, containing the platinum boat, was placed into a tube furnace which had been heated to about 900° C., and was heated for about 3 minutes in flowing oxygen.
- 8. The sample was then furnace-cooled to room temperature in flowing oxygen, the sample was then taken out of the furnace.

The samples prepared by this procedure formed a layer of superconductive compounds on their bottom and a zero resistance temperature of above 100 k. FIG. 1 illustrates the resistance as a function of temperature for a Tl₂O₃-vapor-processed Ba₂Ca₂Cu₃O₇ sample made pursuant to this example. This sample reaches 15 zero resistance at 104 k. FIG. 4 illustrates comparable behavior for a Tl₂O₃ vapor-processed Ba₂Ca₂Cu₃O₇ thick wire precursor made by a similar procedure.

EXAMPLE 2

- A. The following reagents were utilized:
 - 1. Tl₂O₃,
 - 2. CaO,
 - 3. BaCO₃,
 - 4. CuO.
 - B. The following procedure was followed:
 - 1. A mixture of a one molar portion of BaCO₃, a three molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black BaCa₃Cu₃O₇ powder.
 - 2. The resulting BaCa₃Cu₃O₇ powder was completely ground, and pressed into a pellet.
 - The pellet was heated at approximately 925° C. in a tube furnace for about 10 minutes.
 - 4. The heated pellet was then taken out of the furnace and cooled in air to room temperature.
 - 5. A small amount (approximately 0.1 to about 0.2 gm) of Tl₂O₃ was placed in a platinum boat, and the platinum boat was then placed in a quartz boat.
 - 6. The cooled pellet was placed over the platinum
 - 7. The quartz boat was put into the tube furnace which had been heated to about 900° C., and was heated for about 3 minutes in flowing oxygen.
 - 8. The sample was then furnace-cooled to room temperature in flowing oxygen, and the sample was then taken out of the furnace.

The samples prepared by this procedure formed a 50 layer of superconductive compounds on their bottom surface, which had an onset temperature of above 120 K, and a zero resistance temperature of above 100 K. FIG. 2 illustrates resistance as a function of temperature for a Tl₂O₃-vapor-processed BaCa₃Cu₃O₇ sample prepared pursuant to this example. This sample reaches zero resistance at 110 K. FIG. 3 illustrates comparable behavior for a Ba₂CaCu₂O₅ precursor similarly pre-

EXAMPLE 3

- A. The following reagents were utilized:
 - 1. Tl₂O₃,
 - 2. CaCO₃,
 - BaCO₃
 - 4 CuO.
- B. The following procedure was followed:
- 1. A mixture of a two molar portion of BaCO₃, a two molar portion of CaCO₃, and a three molar portion

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of CuO was ground with an agate mortar and pestle, heated in air at approximately 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black Ba₂Ca₂Cu₃O₇ powder.

- 2. The resulting Ba₂Ca₂Cu₃O₇ powder was completely ground, and pressed into a pellet.
- The pellet was heated at approximately 925° C. in a tube furnace for about 10 minutes.
- 4. The heated pellet was then taken out of the furnace and cooled in air to room temperature.
- A small amount (approximately 0.1 to about 0.2 gm) of Tl₂O₃ was put in a platinum boat, and the platinum boat was placed in a quartz boat.
- The cooled pellet was placed over the platinum boat.
- The quartz boat was put into the tube furnace which had been heated to approximately 900° C., and was heated for about 3 minutes in flowing oxygen.
- 8. The sample was then furnace-cooled to room temperature in flowing oxygen, and was then removed from the furnace.

The samples prepared by this procedure formed a layer of superconductive compounds on their bottom surface, which have an onset temperature of above 120 25 K, a midpoint of about 110 K, and a zero resistance temperature of above 100 K.

EXAMPLE 4

- A. The following reagents were utilized:
 - 1. Tl₂O₃,
 - 2. CaO,
 - 3. BaCO₃,
 - 4. CuO.
- B. The following procedure was followed:
- A mixture of a two molar portion of BaCO₃, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain a 40 uniform black Ba₂Ca₂Cu₃O₇ powder.
- The resulting Ba₂Ca₂Cu₃ powder was placed on a
 platinum substrate which was put in a quartz boat,
 and was heated in a tube furnace at approximately
 950 to about 1000° C. for 3-5 minutes in flowing 45
 oxygen.
- The quartz boat was then taken out of the furnace.
 The Ba₂Ca₂Cu₃O₇ powder had melted completely, forming a layer of recrystalline Ba-Ca-Cu-O.
- 4. A small amount (approximately 0.1 to about 0.2 50 gm) of Tl₂O₃ was put in a platinum boat, and the platinum boat was placed in a quartz boat.
- The platinum substrate was put over the platinum boat with the molten Ba-Ca-Cu-O facing downward.
- 6. The quartz boat was put into the tube furnace which had been heated to about 900° C., and was heated for about 3 minutes in flowing oxygen.
- The sample was then furnace-cooled to room temperature in flowing oxygen, and was then taken out 60 of the furnace.

A Tl₂O₃-vapor-processed Ba₂Cu₃O₇ thick film was produced that was superconducting. FIG. 5 illustrates resistance as a function of temperature for a Tl₂O₃-vapor-processed Ba₂Ca₂Cu₃O₇ recrystallized thick film 65 created pursuant to this example. The film had an onset temperature of above 120 K and reached zero resistance at about 111 K.

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This experiment also shows that thin film Tl-Ca-Ba-Cu-O superconductors can be made using the Tl₂O₃ vapor process with appropriately deposited Ca-Ba-Cu-O precursor thin films. These thin films can be prosuced by depositing a thin-film of Ca-Ba-Cu-O precursor utilizing known techniques of physical or chemical deposition. These techniques include, inter alia, sputtering, evaporation, ablation, electrodeposition, electroless deposition, and chemical vapor deposition. After the 10 thin film of Ca-Ba-Cu-O is produced, it can then be reacted with Tl₂O₃ vapor, for example, by placing Tl₂O₃ in a boat under or near the precursor and heating the boat.

EXAMPLE 5

- A. The following reagents were utilized:
- 1. Tl₂O₃,
 - 2. In₂O₃,
 - 3. CaO,
 - 4. BaCO₃,
 - 5. CuO.

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- B. The following procedure was followed:
- A mixture of two molar portion of BaCO₃, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at approximately 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black Ba₂Ca₂Cu₃O₇ powder.
- A one molar portion of the resulting Ba₂Ca₂Cu₃O₇
 powder was mixed with a one molar portion of
 In₂O₃, and was completely ground and pressed into
 a pellet.
- 3. The pellet was heated in a tube furnace at about 900° C. in flowing oxygen for about 3 minutes, and was then removed from the furnace.
- 4. A small amount (approximately 0.1 to about 0.2 gm) of Tl₂O₃ was put in a platinum boat, and the platinum boat was put in a quartz boat
- 5. The In₂Ba₂Ca₂Cu₃O₇ pellet was put over the platinum hoat
- 6. The quartz boat was then put into the tube furnace which had been heated to approximately 90° C., and was heated for about 3 minutes in flowing oxygen.
- The sample was then furnace-cooled to room temperature in flowing oxygen, and was then removed from the furnace.

The bottom surface of the Tl₂O₃-vapor-processed In₂Ba₂Ca₂Cu₃O₇ sample constructed pursuant to this method was found to be superconducting FIG. 6 illustrates resistance as a function of temperature for a Tl₂O₃-vapor-processed In₂Ba₂Ca₂Cu₃O₇ sample, made pursuant to this example, which had an onset temperature about 120 K, and reached zero resistance at 89 K.

EXAMPLE 6

- A. The following reagents were utilized:
 - 1. Tl₂O₃,
 - 2. CaO,
- 3. BaCO₃,
- 4. CuO.
- B. The following procedure was followed:
- A mixture of a two molar portion of BaCO₃, a two molar portion of CaO, and a three molar portion of CuO was ground with an mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black Ba₂Ca₂Bu₃O₇ powder.

2. The resulting Ba₂Ca₂Cu₃O₇ powder was completely ground, and pressed into a pellet.

3. The pellet was heated in a tube furnace at approximately 925° C. for about 5 minutes.

- 4. The pellet was then taken out of the furnace and 5 cooled in air to room temperature.
- 5. A small amount (approximately 0.1 to about 0.2 gm) of Tl₂O₃ was put in a platinum boat, and the platinum boat and cooled pellet were put in a gold container.
- 6. The gold container, was sealed so that the platinum boat and pellet were sealed in oxygen, was put into a tube furnace which had been heated to approximately 900° C., and was heated for about 10 min-
- 7. The gold container was then furnace-cooled to room temperature, and was removed from the furnace.
- 8. The gold container was opened, and the sample $_{20}$
- This example produced a Tl₂O₃-vapor-processed sample that formed a layer of superconducting compounds on its surface, which had an onset temperature of above 120 K and a zero resistance tem- 25 perature of above 100 K.

EXAMPLE 7

- A. The following reagents were utilized: 1. Tl₂O₃,
- 2. CaO,
- 3. BaCO₃,
- 4. CuO.
- B. The following procedure was followed:
- A mixture of a two molar portion of BaCO₃, a two 35 Tl-223. molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at approximately 925° C. (with several intermediate grindings) to obtain a uniform black Ba₂Ca₂Cu₃O₇ powder.
- 2. The resulting Ba₂Ca₂Cu₃O₇ powder was mixed with Tl₂O₃ in a molar ratio of 1:1, and was completely ground and pressed into a pellet.
- 3 The pellet was heated at approximately 925° C. in a tube furnace for approximately 4 hours. The sam- 45 ple could have been heated for a longer time however.
- 4. The heated pellet was then taken out of the furnace and cooled in air to room temperature. The resultant pellet was found to have a semiconductor behavior.
- 5. A small amount (approximately 0.1 to about 0.2 gm) of Tl₂O₃ was put in a platinum boat, and the platinum boat was put in a quartz boat.
- 6. The resultant pellet was put over the platinum
- 7. The quartz boat was then put into the tube furnace which had been heated to approximately 900° C., and was heated for about 3 minutes in flowing 60
- 8. The sample was then furnace-cooled to room temperature in flowing oxygen, and was then removed from the furnace.

The resulting Tl₂O₃-vapor-processed sample, pro- 65 duced by this example, had a superconducting behavior, and had an onset temperature of above 120 K, and a zero resistance temperature of above 100 K.

EXAMPLE 8

- A. The following reagents were utilized:
 - 1. Tl₂O₃,
 - 2. CaO,
 - 3. BaCO₃,
 - 4. CuO.
- B. The following procedure was followed:
- 1. A mixture of a two molar portion of BaCO₃, a two molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain a uniform black Ba₂Ca₂Cu₃O₇ powder. A pellet was then formed in a \{\} inch die under pressure.
- 2. A Ba₂Ca₂Cu₃O₇ precursor film was then deposited by laser deposition from the pellet as a target onto a Y-stabilized ZrO2 substrate using a frequencydoubled Nd-YAG laser operating at 523 nm, forming a Ba-Ca-Cu-O thin film with a thickness of approximately 2 to about 3 μ m.
- 3. The thin film precursor was then placed above a platinum boat which contained approximately 0.1 to about 0.2 gram of Tl₂O₃, and the platinum boat was placed in a quartz boat.
- 4. The quartz boat and its contents was heated at 900 C. in a tube furnace for approximately 2 to about 3 minutes in flowing oxygen, and then furnace cooled to room temperature.
- A resultant Tl₂O₃ vapor processed Ba₂Ca₂Cu₃O₇ thin film was found to be superconducting with an onset temperature above 120 K and a zero resistance temperature of 115 K. The measurements of a sample prepared pursuant to this example are illustrated in FIG. 7 as

EXAMPLE 9

- A. The following reagents were utilized:
 - 1. Tl₂O₃,
 - 2. CaO,
 - 3. BaCO₃,
 - CuO.
- B. The following procedure was followed:
- 1. A mixture of a two molar portion of BaCO₃, a three molar portion of CaO, and a three molar portion of CuO was ground with an agate mortar and pestle, heated in air at 925° C. for more than 24 hours (with several intermediate grindings) to obtain an uniform black Ba₂Ca₃Cu₃O₈ powder. A pellet was then formed in a 3 inch die under pressure.
- 2. A Ba₂Ca₃Cu₃O₈ precursor was deposited by laser deposition onto a Y-stabilized ZrO2 substrate using a frequency-doubled Nd-YAG laser operating at 523 nm, forming a Ba-Ca-Cu-O thin film with a thickness of approximately 2 to about 3 µm.
- 3. The thin film precursor was placed above a platinum boat which contained approximately 0.1 to about 0.2 gram of Tl₂O₃, and the platinum boat was placed in a quartz boat.
- 4. The quartz boat and its contents was heated at 900 C. in a tube furnace for approximately 2 to about 3 minutes in flowing oxygen, and then furnace cooled to room temperature.

A resultant Tl₂O₃ vapor processed Ba₂Ca₃Cu₃O₈ thin film was found to be superconducting with an onset temperature above 120 K and a zero resistance temperature of 104 K. The measurements of a sample prepared pursuant to this method are illustrated in FIG. 7 as TI-233.

It should be understood that various changes and modifications to the presently preferred embodiments described here in be apparent to those skilled in the art. 5 Such changes and modifications can be made without departing from the spirit and scope of the present invention and without diminishing its attendant advantages. It is therefore intended that such changes and modifications be covered by the appended claims.

We claim:

- 1. A method for making thin-film Tl-Ca-Ba-Cu-O superconductors comprising the steps of:
 - a. depositing a thin-film of Ca-Ba-Cu-O precursor utilizing physical vapor deposition;
 - b. placing Tl₂O₃ in a boat under or near the precursor thin film;
 - c. heating the boat and precursor film in flowing O₂;
 and
 - d. cooling the film and boat.
- 2. A method for making thin-film Tl-Ca-Ba-Cu-O superconductors comprising the steps of:

mixing and heating BaCO₃, CaO, and CuO to obtain a Ca-Ba-Cu-O powder;

forming a Ca-Ba-Cu-O thin film from the Ca-Ba-Cu-O powder by physical or chemical deposition; and heating the film in oxygen with a quantity of Tl₂O₃ in oxygen to create a Tl₂O₃ vapor processed film.

- 3. The method of claim 2 wherein the powder includes the composition Ba₂Ca₂Cu₃O₇.
- 4. The method of claim 2 wherein the powder includes the composition Ba₂Ca₃Cu₃O₈.
- 5. The method of claim 2 wherein two molar portions of BaCO₃, two molar portions of CaO, and three molar portions of CuO are ground, mixed, and heated to cre10 ate the powder.
 - 6. The method of claim 2 wherein two molar portions of BaCO₃, three molar portions of CaO, and three molar portions of CuO are ground, mixed, and heated to create the powder.
 - 7. The method of claim 2 wherein the powder is created into a pellet.
 - 8. The method of claim 7 wherein a film is deposited by laser deposition from the pallet into an Y-stabilized ZrO₂ substrate.
 - 9. The method of claim 8 wherein the film has a thickness of approximately 2 to about 3 μ m.
 - 10. The method of claim 2 wherein the Tl₂O₃ is placed in a platinum boat and the film is placed above the boat prior to heating the film and Tl₂O₃.
 - 11. The method of claim 2 wherein the quantity of Tl₂O₃ is approximately 0.1 to about 0.2 g.

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