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Melt-produced high temperature superconductor

Allen M. Hermann University of Arkansas, Fayetteville

Zhengzhi Sheng University of Arkansas, Fayetteville

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[54] MELT-PRODUCED HIGH TEMPERATURE RARE EARTH BARIUM COPPER OXIDE SUPERCONDUCTOR AND PROCESSES FOR MAKING SAME

[75] Inventors: Allen M. Hermann; Zhengzhi Sheng, both of Fayetteville, Ark.

[73] Assignee: University of Arkansas, Fayetteville,

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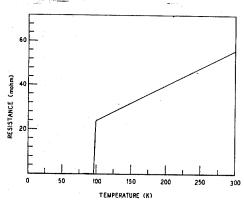
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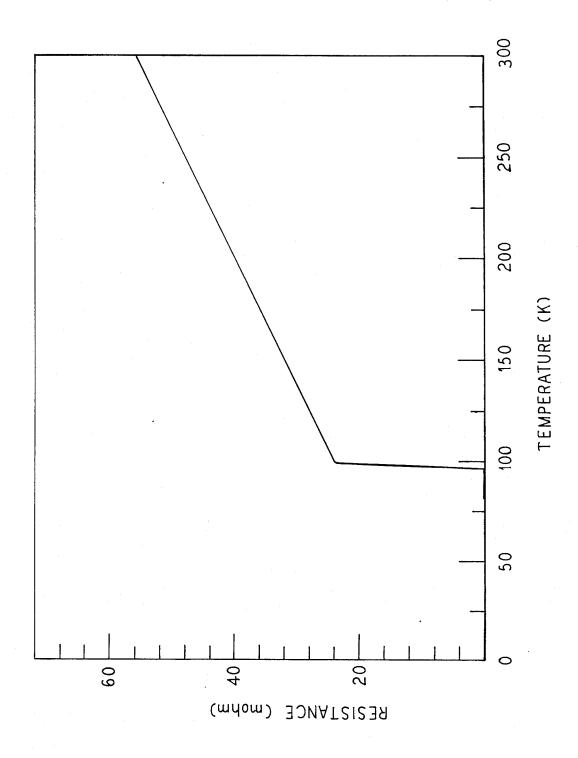
Primary Examiner—Dennis Albrecht Attorney, Agent, or Firm—Hill, Van Santen, Steadman & Simpson

[57] ABSTRACT

A class of melt-produced, high temperature suerpconductors and processes of making same are provided. The superconductor has a preferred composition of R-Ba-Cu-O wherein R is chosen from the group of rare earth metals excluding: Praseodyium; Cerium; and Terbium. The process is carried out at a relatively low temperature of about 950° C., and the process allows fabrication of melt-produced high temperature superconductors of arbitrary shape. The process is based on the reaction between molten barium-copper oxides and solid rare earth oxides, rare earth barium oxides, rare earth copper oxides, or rare earth barium-copper oxides. In an embodiment, the method comprises the steps of: mixing and grinding BaCO3 and CuO with other nominal compositions; pressing the resultant mixture into a pellet, if necessary; placing the pellet or powder on a pellet or powder that can include rare earth copper oxides; heating the pellet and/or powders to a temperature of approximately 950° C.; and removing a meltproduced superconductor from the remaining powder or pellet.

35 Claims, 1 Drawing Sheet





MELT-PRODUCED HIGH TEMPERATURE RARE EARTH BARIUM COPPER OXIDE SUPERCONDUCTOR AND PROCESSES FOR MAKING SAME

BACKGROUND OF THE INVENTION

The present invention relates generally to high temperature superconductors. More specifically, the present invention relates to a melt-produced, high temperature superconductor and processes for making same.

Recently, there has been much work done on the utilization of ternary oxides containing copper, barium, and rare earth elements for superconductivity at temperatures above 90° K. with the belief that superconductivity at or above room temperature is possible. Some of this work has centered on the utilization of a yttrium (Y)-barium (Ba)-copper (Cu)-oxygen (O) system.

Current developments of the Y-Ba-Cu-O class of ceramic superconductors have yielded optimistic results. These systems are typically created utilizing bulk materials which are powders or very small granule substances. These powders must be compressed to facilitate measurements and are relatively intractable. As can be appreciated, the structure of these substances is not conductive to the manufacturing of components from these ceramic superconductors.

A melt-produced, high temperature superconductor 30 would be desirable for many reasons. Melt-produced superconductors would: (1) allow the growth of large bulk single crystals—this is important because it could facilitate the discovery of the correct theory on superconductivity; (2) provide low cost processing and 35 manufacturability; (3) allow components to be cast in any shape; (4) allow the addition of other components to the melt-produced high temperature superconductors—this would allow one to create an extrudeable composition allowing the superconductor to be useful 40 in the manufacture of superconducting wires, magnets, etc.; and (5) the inventors believe, allow for high critical currents in the high temperature superconductor allowing the generation of large currents therethrough.

Accordingly, there is a need for melt-produced high 45 temperature superconductors and processes for making same.

SUMMARY OF THE INVENTION

The present invention provides melt-produced high 50 temperature superconductors and processes for making same that are unique to date among high temperature superconductors.

The present invention preferably comprises a composition having the following formula:

R-Ba-Cu-O

wherein:

R is chosen from the group of rare earth metals excluding: Praseodymium (Pr); Cerium (Ce); and Terbium (Tb).

Preferably, R is chosen from the group of rare earth metals that include: Yttrium (Y); Gadolinium (Gd); Erbium (Er); Holmium (Ho); Neodymium (Nd); 65 Samarium (Sm); Europium (Eu); Ytterbium (Yb); Dysprosium (Dy); Thulium (Tm); and Lutetium (Lu). In a preferred embodiment, R is Yttrium.

In a preferred embodiment, the melt-produced superconductor has the following nominal stoichiometry:

RBa₂Cu₃O_{6.5}

wherein:

R is a rare earth metal not including: Pr; Tb; or Ce. Preferably, R is chosen from the group of rare earth metals that include: Yttrium (Y); Gadolinium (Gd); Erbium (Er); Holmium (Ho); Neodymium (Nd); Samarium (Sm); Europium (Eu); Ytterbium (Yb); Dysprosium (Dy); Thulium (Tm); and Lutetium (Lu).

In a preferred embodiment, the melt-produced super-15 conductors has the following nominal stoichiometry:

YBa2Cu3O6.5.

A method of making the melt-produced, high temperature superconductors is also provided. The method is based on a reaction between molten Ba-Cu oxides and solid rare earth oxides, rare earth barium oxides, rare earth copper oxides, or rare earth-Ba-Cu oxides. The method allows melt-producing to be carried out at a relatively low temperature of approximately 950° C. The method further allows the use of precursor materials for the superconductor in the form of pellets, powders, or pellets and powders.

The method comprises the steps of: mixing and grinding BaCO₃ and CuO with nominal compositions of Ba₂Cu₃O₅, BaCu₃O₄, BaCu₄O₅, BaCu₆O₇, and BaCu₁. 2O₁₃; pressing the resultant mixture into a pellet, if necessary; placing the pellet or powder on a pellet or powder chosen from the group consisting of R₂O₃, or RBa-oxides (e.g. RBaO_{2.5}), R-Cu-oxides, or R-Ba-Cu-oxides (e.g. R_{1.2}Ba_{0.8}CuO_{3.6}) wherein R is chosen from the group of rare earth metals excluding: Tb, Pr, and Ce; heating the pellets and/or powders to a temperature of approximately 950° C.; and removing a melt-produced superconductor from the remaining pellet or powder at the bottom.

In an embodiment of the method, a mixture of BaCO₃ and CuO with nominal composition of BaCu-3O4 is ground and heated to approximately 900° C. in air for about 12 hours. The heated mixture is reground and pressed into pellets. A mixture of Y2O3, BaCO3, and CuO with nominal composition of Y_{1.2}Ba_{0.8}CuO_{3.6} is ground and heated at approximately 950° C. in air for about 12 hours. The resultant material is reground. The black BaCu₃O₄ pellet (of arbitrary shape) is placed on the (green) Y_{1.2}Ba_{0.8}CuO_{3.6} powder, and is heated in flowing O₂ at about 950° C. for 12 hours, then at 650° C. for 4 hours, then cooled down to 200° C. in about 1-2 hours, then removed from the furnace. A black meltproduced superconductor is formed which is embedded in the (green) Y_{1.2}Ba_{0.8}CuO_{3.6} powder and separated from it (the recovered Y_{1,2}Ba_{0.8}CuO_{3.6} powder may be reused if desired).

Accordingly, an advantage of the present invention is to provide a melt-produced, high temperature superconductor and method of making same.

A further advantage of the present invention is to provide a method that affords the ability to grow large single crystals of this supeconductor.

A still further advantage of the present invention is that it provides a melt-produced, high temperature superconductor that is easily formable and manufactur-

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able allowing it to be utilized in commercial applications.

Furthermore, an advantage of the present invention is that it provides a melt-produced, high temperature superconductor to which other components can be added 5 to make, for example, wire-type high temperature superconductors.

Still another advantage of the present invention is that it provides a high temperature superconductor having high critical currents.

Moreover, an advantage of the present invention is that it provides a high temperature superconductor that can be used to carry high currents without energy loss.

A further advantage of the present invention is that it provides a method for making melt-produced, high 15 temperature superconductors at relatively low temperatures of about 950° C.

A still further advantage of the present invention is that it provides a supeconductor that can be utilized to levitate vehicles, store energy in magnetic fields, and 20produce more intense magnetic fields than heretofore possible.

Additional advantages and features of the present invention are described in, and will be apparent from, bodiments.

BRIEF DESCRIPTION OF THE DRAWING

The FIG. illustrates the resistance of a sample of a superconductor made by the method of the present 30 invention as a function of temperature.

DETAILED DESCRIPTION OF THE PRESENTLY PREFERRED EMBODIMENTS

The present invention provides a melt-produced, 35 high temperature superconductor and process for making same. As used herein, the term "high temperature" refers to a temperature above the boiling temperature of

It has been found that the melt-produced, high tem- 40 perature supeconductor of the present invention upon reaction of the molten and solid constituents is superconducting with a critical temperature of at least 90° K.

The melt-produced, high temperature superconductor composition of the present invention preferably has 45 the formula:

R-Ba-Cu-O

R is chosen from the group of rare earth metals excluding: Praseodymium (Pr); Terbium (Tb); and Cerium (Ce).

Preferably, R is a rare earth metal chosen from the group consisting of: Yttrium (Y); Gadolinium (Gd); 55 Erbirum (Er); Holmium (Ho); Neodymium (Nd); Samarium (Sm); Europium (Eu); Ytterbium (Yb); Dysoprosium (Dy); Thulium (Tm); and Lutetium (Lu).

In a preferred embodiment, R is Yttrium.

Preferably, the melt-produced, high temperature su- 60 perconductor has the following nominal stoichiometry:

RBa₂Cu₃O_{6.5}

wherein:

R is a rare earth metal exlucing: Tb; Pr; and Ce and preferably: Y; Gd; Er; Ho; Nd; Sm; Eu; Yb; Dy; Tm: and Lu.

In a preferred embodiment, the melt produced, high temperature supeconductor has the following nominal stoichiometry:

YBa2Cu3O6.5.

The present invention provides methods for making meltproduced high temperature superconductors at temperatures as low as 950° C.

By way of example, and not limitation, examples of the methods of making the melt-produced, high temperature superconductors will now be given.

EXAMPLE 1

A. Y₂O₃, BaCO₃, and CuO with nominal compositions of Ba₂Cu₃O₅ and Y_{1,2}Ba_{0,8}CuO_{3,6} were mixed and ground in an agate mortar. The powders were heated to approximately 900° C. in air for approximately 12 to about 24 hours.

B. The heated mixtures were then reground. The resultant Ba₂Cu₃O₅ powder was then pressed into a pellet. The Ba₂Cu₃O₅ pellet was then placed on the Y_{1.2}Ba_{0.8}CuO_{3.6} powder. The pellet and powder were the detailed description of the presently preferred em- 25 then heated at approximately 950° C. in flowing oxvgen, for approximately 24 hours in a tube furnace. The tube furnace was then cooled to approximately 650° C. and remained at that temperature for approximately 2 to about 4 hours. The tube furnace was then cooled to approximately 200° C. before the pellet and powder were removed from the furnace.

> C. During the heating process, part of the Ba₂Cu₃O₅ pellet method into the Y_{1.2}Ba_{0.8}CuO_{3.6} powder. This melt-produced material was black in color and was superconductive.

The figure illustrates resistance (as a function of temperature) of the melt-produced superconductor embedded in the powder. It is seen that the superconductor has a high transition temperature. Specifically, the midpoint transition temperature is 93° K and the transition width is about 2° K.

The superconductive material had a high density of approximately 6.4 grams per cm³. X-ray diffraction showed that the superconductor consisted of YBa2. Cu₃O_{6.5} (nominal composition) with very little CuO as an impurity. The melt-produced, superconductor demonstrated a strong Meissner effect and levitated in a magnetic field of 8 kiloGauss.

EXAMPLE 2

An experiment was performed to find a Ba-Cu-oxide system with a low melt point. It should be noted that in the previous example, at 950° C., more than half of Ba₂Cu₃O₅ did not melt. In order to search for a Ba-Cuoxide system with a lower melting point, the following experiment was performed.

A. CuO or mixtures of BaCO₃ and CuO with nominal compositions of Ba₃Cu₂O₅, Ba₂Cu₃O₅, BaCu₃O₄, BaCu₄O₅, BaCu₆O₇, BaCU₁₂O₁₃, and BaCu₄₈O₄₉, were weighed and ground in an agate mortar.

B. The resultant powders were heated in air at 950° C. for approximately 8 hours.

The experiment demonstrated that BaCU₃O₄ and BaCU₄O₅ are the Ba-Cu-oxides with the lowest melting points. To this end, these powders were essentially completely melted at 950° C.

EXAMPLE 3

A. A mixture of BaCO₃ and CuO with a nominal composition of BaCU₃O₄ was ground in an agate mortar, and pressed into a pellet.

B. The resultant pellet was put on a powder consisting of Y_{1,2}Ba_{0,8}CuO_{3,6} (nominal composition) that had been heated at approximately 950° C. The pellet and powder were then heated at approximately 950° C. in flowing O2 for approximately 24 hours in a tube fur- 10 nace. The furnace was then cooled to approximately 650° C. and remained at 650° C. for about 2 to about 4 hours. The furnace was then turned off, and, when the furnace temperature was below 200° C., the pellet and powder were removed from the furnace.

After removing the heated materials from the furnace it was noted that the BaCu₃O₄ pellet had almost completely melted into the powder. A black chunk of superconductor material was formed, embedded in the remaining Y_{1.2}Ba_{0.8}CuO_{3.6} powder that was green in 20 color. This superconductor had properties similar to those of Example No. 1.

EXAMPLE 4

A. A mixture of BaCO₃ and CuO with a nominal 25 composition of BaCU₃O₄ was ground in an agate mortar. The resultant powder was then pressed into a pellet.

B. A mixture of Y₂O₃ and CuO with a nominal com-

position YBaO_{2.5} was also ground.

C. The BaCu₃O₄ pellet was put on the resultant 30 YBaO_{2.5} powder of Step B and heated at 950° C. in flowing O2 for approximately 24 hours in a tube furnace. The temperature of the furnace was then reduced to 650° C. and the pellet and powder were left in the furnace at this temperature for approximately 2 to about 35 4 hours. The furnace was then cooled to approximately 200° C. before the pellet and powder were removed from the furnace.

The BaCu₃O₄ pellet was almost completely melted into the powder. This resulted in a black chunk of su- 40 perconductor being formed, that was embedded in the remaining white YBaO_{2.5} powder. This superconductor had properties similar to those of Example No. 1.

EXAMPLE 5

A. A Ba₂Cu₃O₅ pellet was put on a powder that consisted of Y₂O₃. The pellet and powder were heated at 950° C. in flowing O₂ for approximately 12 to about 24 hours in a tube furnace. The temperature of the furnace was reduced to 650° C. and the pellet and powder re- 50 mained therein for approximately 2 to about 4 hours. The furnace was then cooled to 200° C. before the pellet and powder were removed from the furnace.

A part of the Ba₂Cu₃O₅ pellet was melted. The interface between the Ba₂Cu₃O₅ pellet and Y₂O₃ powder 55 contained a black piece of melt-produced material. The material was found to have superconductive properties.

EXAMPLE 6

A. A Ba₂Cu₃O₅ pellet was placed on YBa₂Cu₃O_{4.5}F₄ 60 powder which had been heated to approximately 950° C. The pellet and powder were heated to approximately 950° C. in flowing O₂ for approximately 24 hours in a tube furnace. The furnace was cooled to 650° C. and the pellet and powder remained therein for approximately 2 65 to about 4 hours. The furnace was cooled to 200° C. before the pellet and powder were removed from the furnace.

A part of the Ba₂Cu₃O₅ pellet was melted into the powder. A black piece of superconductor was formed in the interface of the pellet and the powder. This superconductor had properties similar to those of Example 5 No. 1.

EXAMPLE 7

In this example, superconductors were created from a Ba₂Cu₃O₅ pellet melted into a powder with nominal composition R_{1.2}Ba_{0.8}CuO_{3.6} wherein R is a rare earth chosen from the group consisting of: Nd; Sm; Eu; and Dy.

A. BaCO₃ and CuO with a nominal composition of Ba₂Cu₃O₅ were ground and heated in air at approximately 900° C. for 12 hours. The mixture was then reground and pressed into pellets.

B. R₂O₃ (wherein R is chosen from the group consisting of: Nd; Sm; Eu; and Dy), BaCO3 and CuO with a nominal composition of R_{1.2}Ba_{0.8}CuO_{3.6} were ground and heted in air at approximately 950° C. for 12 hours and then cooled down to less than 200° C., and reground.

C. A Ba₂Cu₃O₅ pellet was put on the resultant R_{1,2}Ba_{0.8}CuO_{3.6} powder, and heated in O₂ at approximately 950° C. for approximately 15 hours in a tube furnace. The furnace was then cooled to approximately 650° C. and the pellet and powder were heated therein for approximately 2 hours. The furnace was then cooled down to 200° C. and the pellet and powder removed.

The Ba₂Cu₃O₅ pellet was partly melted, and a black chunk of melt-produced superconductor was formed, which embedded in the remaining R_{1.2}Ba_{0.8}CuO_{3.6} powder. The formed superconductor exhibited properties similar to those exhibited by the superconductors produced in Example No. 1. The superconductors had the following nominal stoichiometry: RBa₂Cu₃O_{6.5}, wherein RNd; Sm; Eu; or Dy.

The inventors believe that R=Gd, Ho, Er, Tm, Yb, and Lu would also work to create a superconductor following the steps of the above example.

EXAMPLE 8

A. R₂O₃ (wherein R is chosen from the group consist-45 ing of: Gd; Ho; Er; and Yb) and BaCO3 with nominal composition of RBaO_{2.5} were ground and heated in air at approximately 950° C. for approximately 6 hours and

B. BaCO₃ and CuO with a nominal composition of BaCu₃O₄ were ground and pressed into pellets.

C. A BaCu₃O₄ pellet was put on an RBaO_{2.5} powder and heated in O2 at approximately 950° C. for approximately 5 hours. The tube furnace was then cooled down to less than 200° C. before the materials were removed from the furnace.

The BaCu₃O₄ pellet was almost completely melted, and a black chunk was formed, which was embedded in the remaining RBaO_{2.5} powder. The superconductor had properties similar to the superconductors produced in Example No. 1.

The inventors believe a melt-produced superconductor can also be produced if R=Nd, Sm, Eu, Dy, Tm and Lu and the above steps are followed. EXAMPLE 9

A. Ho₂O₃ and BaCO₃ with nominal composition of HoBa₂O_{3.5} were ground and heated in air at approximately 950° C. for approximately 6 hours. The mixture was then reground.

B. BaCO₃ and CuO with a nominal composition of BaCu₃O₄ were ground and pressed into pellets.

C. The resultant BaCu₃O₄ pellet was put on HoBa₂O_{3.5} powder, heated in O₂ at approximately 950° C. for approximately 5 hours in a tube furnace. The furnace 5 was cooled down to less than 200° C. before the materials were removed from the furnace.

The BaCu₃O₄ pellet had almost completely melted, and a black chunk was formed, which was embedded in the remaining HoBa₂O_{3.5} powder. The black chunk had 10 superconductive properties similar to those of the previous examples.

EXAMPLE 10

A. Ho₂O₃ and SrCO₃ with nominal composition of 15 HoSrO_{2.5} were ground and heated in air at 950° C. for 6 hours and reground.

B. BaCO₃ and CuO with a nominal composition of BaCU₃O₄ were ground and pressed into pellets.

C. The resultant BaCu₃O₄ pellet was put on HoS- 20 rO_{2.5} powder, heated in O₂ at approximately 950° C. for approximately 5 hours in a tube furnace. The furnace was then cooled down to less than 200° before the pellet and powder were moved out of the furnace.

The BaCu₃O₄ pellet was almost completely melted, 25 and a black chunk was formed, which was embedded in the remaining HoSrO_{2.5} powder. The black chunk had superconductive properties similar to those of the previous examples.

EXAMPLE 11

The present example deals with the fabrication of arbitrarily-shaped components made from melt-produced superconductors. The example given is for a ring shape.

A. BaCO₃ and CuO with nominal composition of BaCu₃O₄ were mixed and ground, and then heated in air at 900° C. for 12 hours. The resultant black material was reground and pressed into pellets (hereafter called Ba-Cu pellets).

B. Y₂O₃, BaCO₃, and CuO were mixed to a nominal composition of Y_{1.2}Ba_{0.8}CuO_{3.6} and ground, and then heated in air at 950° C. for 12 hours. The resultant (green) material was then powdered (hereafter called Y_{1.2} powder).

C. The $Y_{1,2}$ powder was packed into a crucible and the surface leveled. Ba-Cu pellets were then placed one-by-one onto the $Y_{1,2}$ powder to form a ring shape.

D. The crucible and its contents were then heated in a tube furnace in flowing O₂ at 950° C. for about 12 50 hours, then held at 650° C. for 4 hours, and then cooled down to below 200° C., at which point they were removed from the furnace and cooled to room temperature

E. A black melt-produced superconductive ring was 55 formed, embedded in the $Y_{1.2}$ (green) powder. The ring was removed from the powder. If desired, the powder could then be reground and used again in similar fashion.

EXAMPLE 12

The present example deals with the fabrication of arbitrarily-shaped components made from melt-produced superconductors.

A. BaCO₃ and CuO with nominal composition of 65 BaCu₃O₄ were mixed and ground, and then heated in air at 900° C. for 12 hours. The resultant black material was reground.

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B. Y_2O_3 , BaCO₃, and CuO were mixed to a nominal composition of $Y_{1.2}Ba_{0.8}CuO_{3.6}$ and ground, and then heated in air at 950° C. for 12 hours. The resultant (green) material was then powdered (hereafter called $Y_{1.2}$ powder).

C. The $Y_{1,2}$ powder was packed into a crucible and the surface leveled. BaCu₃O₄ powder was then placed onto the $Y_{1,2}$ powder.

D. The crucible and its contents were then heated in a tube furnace in flowing O₂ at 950° C. for about 12 hours, then held at 650° C. for 4 hours, and then cooled down to below 200° C., at which point they were removed from the furnace and cooled to room temperature.

E. A black melt-produced superconductive compound was formed, embedded in the Y_{1.2} (green) powder. The compound was removed from the powder. The powder could then be reground and used again in similar fashion.

EXAMPLE 13

The present example deals with the fabrication of arbitrarily-shaped components made from melt-produced superconductors. The example given is for a ring shape.

A. BaCO₃ and CuO with nominal composition of BaCu₃O₄ were mixed and ground, and then heated in air at 900° C. for 12 hours. The resultant black material was reground and pressed into a ring (hereafter called BaCu₃O₄ ring).

G. Y₂O₃, BaCO₃, and CuO were mixed to a nominal composition of Y_{1.2}Ba_{0.8}CuO_{3.6} and ground, and then heated in air at 950° C. for 12 hours. The resultant (green) material was then powdered (hereafter called Y_{1.2} powder).

C. The Y_{1.2} powder was packed into a crucible and the surface leveled. The BaCu₃O₄ ring was then placed onto the Y_{1.2} powder.

D. The crucible and its contents were then heated in a tube furnace in flowing O₂ at 950° C. for about 12 hours, then held at 650° C. for 4 hours, and then cooled down to below 200° C., at which point they were removed from the furnace and cooled to room temperature.

E. A black melt-produced superconductive ring was formed, embedded in the $Y_{1,2}$ (green) powder. The ring was removed from the pocket. The powder could then be reground and used again in similar fashion.

It should be understood that various changes and modifications to the presently preferred embodiments described herein will be apparent to those skilled in the art. Such changes and modifications can be made without departing from the spirit and scope of the present invention and without diminishing its attendant advantages. It is therefore intended that such changes and modifications be covered by the appended claims.

We claim:

1. A method for making a melt-produced high tem-60 perature superconductor comprising the steps of:

mixing and grinding BaCO₃ and CuO with nominal compositions of at least one compound chosen from the group consisting of Ba₂Cu₃O₅, BaCu₃O₄, BaCu₄O₅, BaCu₆O₇, and BaCu₁₂O₁₃;

pressing the resultant mixture into a pellet;

placing the pellet on a powder comprising at least one compound chosen from the group consisting of R_{1.2}Ba_{0.8}CuO_{3.6}, RBa₂Cu₃O_{4.5}F₄, RBa₂O_{3.5}, RBa-

₂O_{2.5}, or R₂O₃ wherein R is chosen from the group of rare earth metals excluding: Tb, Pr, and Ce;

heating the pellet and the powder in a furnace to a temperature of greater than or equal to 950° C. and less than or equal to 1000° C. to form a melt at a 5 pellet: powder interface; and

removing the melt-produced superconductor from the powder in which the superconductor is embedded

- 2. The method of claim 1 wherein R is chosen from ¹⁰ the group of rare earth metals consisting of: Y, Gd, Er, Ho, Nd, Sm, Eu, Yb, Dy, Tm, and Lu.
- 3. The method of claim 1 wherein the oxide chosen is BaCu₃O₄.
- 4. The method of claim 1 wherein the oxide chosen is 15 BaCu₄O₅.
- 5. The method of claim 1 wherein the oxide chosen is $Ba_2Cu_3O_5$.
- 6. The method of claim 1 wherein the pellet and powder are heated in the furnace at a temperature between 940° C. to about 1000° C. for approximately 12 to about 24 hours.
- 7. The method of claim 1 wherein after the pellet and powder are heated to 940°-1000° C. for a predetermined time in the furnace, the temperature of the furnace is lowered to approximately 650° C. and the pellet and mixture remain therein for a predetermined time.
- 8. The method of claim 7 including the step of cooling the furnace to 200° C. or less before removing the pellet and powder therefrom.
- 9. The method of claim 1 wherein after the mixing and grinding step of Ba-Cu oxide, the resultant mixture is heated to approximately 850°-920° C.
- 10. The method of claim 1 wherein the powder on 35 which the Ba-Cu-oxide pellet is placed has been previously heated to approximately 900°-1000° C.
- 11. A method for making a superconductor comprising the steps of:
 - mixing and grinding BaCO₃ and CuO with nominal 40 compositions of at least one compound chosen from the group consisting of Ba₂Cu₃O₅, BaCu₃O₄, BaCu₄O₅, BaCu₆O₇, and BaCu₁₂O₁₃;

heating the resultant mixture;

pressing the resultant mixture into a pellet;

placing the pellet on a preheated powder comprising RBa₂Cu₃O_{4.5}F₄ wherein R is chosen from the group consisting of: Gd, Ho, Er, Yb, Nd, Sm, Eu, Dy, Tm, and Lu;

heating the pellet and the powder in a furnace to 50 approximately 940° to about 1000° C. to form a melt at a powder:powder interface; and

removing a melted interface between the pellet and powder.

12. A method for making a superconductor compris- 55 ing the steps of:

mixing and grinding BaCO₃ and CuO with nominal compositions of at least one compound chosen from the group consisting of Ba₂Cu₃O₅, BaCu₃O₄, BaCu₄O₅, BaCu₆O₇, and BaCu₁₂O₁₃;

heating the resultant mixture;

pressing the resultant mixture into a pellet;

placing the pellet on a powder that has previously been heated comprising R_{1.2}Ba_{0.8}CuO_{3.6} wherein R is chosen from the group consisting of: Y, Gd, Er, 65 Ho, Nd, Sm, Eu, Yb, Dy, Tm, and Lu;

heating the pellet and the powder in a furnace to a temperature of greater than or equal to 950° C. and

less than or equal to 1000° C. to form a melt at a pellet:powder interface; and

removing the melt-produced superconductor from the powder in which the superconductor is embedded.

- 13. The method of claim 12 wherein after the mixing and grinding step, the resultant mixture is heated to approximately 850°-920° C.
- 14. The method of claim 12 wherein the powder the pellet is placed on has been previously heated to approximately 900°-1000° C.
- 15. The method of claim 12 wherein after the pellet and powder are heated to 940°-1000° C. for a predetermined time in the furnace, the temperature of the furnace is lowered to approximately 650° C. and the pellet and mixture remain therein for a predetermined time.
- 16. The method of claim 12 wherein the pellet and powder are heated in the furnace at a temperature between 940° C. to about 1000° C. for approximately 12 to about 24 hours.
- 17. A method for making a superconductor comprising the steps of:

mixing and grinding BaCO₃ and CuO with nominal compositions of at least one compound chosen from the group consisting of BaCu₃O₅, Ba₂Cu₃O₄, BaCu₄O₅, BaCu₆O₇, and BaCu₁₂O₁₃;

heating the resultant mixture to a temperature of approximately 900° C.;

pressing the resultant mixture into a pellet;

placing the pellet on a preheated powder comprising Y_{1.2}Ba_{0.8}CuO_{3.6};

heating the pellet and the powder in a furnace to a temperature of greater than or equal to 940° C. and less than or equal to 1000° C. for approximately 12 to about 24 hours to form a melt at a pellet:powder interface;

cooling the furnace to approximately 650° C. and heating the pellet and powder therein;

cooling the furnace to at least approximately 200° C.; and

removing a melted interface between the pellet and powder.

18. A method for making a superconductor comprising the steps of:

placing a pellet consisting of a compound chosen from the group consisting of Ba₂Cu₃O₅, BaCu₃O₄, and BaCu₄O₅ on a powder consisting of a compound chosen from the group consisting of RBa₂Cu₃O_{4.5}F₄, R_{1.2}Ba_{0.8}CuO_{3.6}, R₂O₃, and RBa₂O_{3.5} wherein R is chosen from the group of rare earth metals excluding Tb, Pr, and Ce;

heating the pellet and powder to approximately 950° C. to about 1000° C. for 12 to 24 hours to form a melt at a pellet:powder interface; and

removing a melted interface between the pellet and powder.

- 19. The method of claim 18 wherein R is chosen from the group of rare earth metals consisting of: Y, Gd, Er, Ho, Nd, Sm, Eu, Yb, Dy, Tm, and Lu.
 - 20. The method of claim 18 wherein R is Y.
- 21. The method of claim 18 wherein after the pellet and powder are heated to 950°-1000° C. for a predetermined time in the furnace, the temperature of the furnace is lowered to approximately 650° C. and the pellet and the mixture remain therein for a predetermined time.

22. The method of claim 18 including the step of cooling the furnace to 200° C. or less before removing the pellet and powder therefrom.

23. The method of claim 18 wherein the pellet and powder are heated in the furnace at a temperature be- 5 tween 950° C. to about 1000° C. for approximately 12 to about 24 hours.

24. The method of claim 18 wherein the powder the pellet is placed on has been previously heated to approximately 900°-1000° C.

25. A method of making a superconductor comprising the steps of:

placing a pellet consisting of BaCu₃O₄ on a powder consisting of HoBa₂O_{3.5};

heating the pellet and powder to approximately 940° 15 C. to 1000° C. to form a melt at a pellet:powder interface; and

removing a melted interface between the pellet and powder.

26. A method of making a superconductor compris- 20 ing the steps of:

placing a pellet consisting of BaCu₃O₄ on a powder consisting of HoSrO_{2.5};

heating the pellet and powder to approximately 950° C. to 1000° C. to form a melt at a pellet:powder 25 interface: and

removing a melted interface between the pellet and powder.

27. A process of making melt-produced superconductor components of arbitrary shape comprising the steps

mixing and grinding BaCO3 and CuO in a nominal composition of BaCu₃O₄;

heating the mixture in air at about 900° C. for about 35 12 hours, cooling the mixture, and grinding it;

mixing and grinding Y2O3, BaCO3, and CuO in a nominal composition of Y_{1,2}Ba_{0,8}CuO_{3,6};

heating the mixture in air at about 950° C. for about 12 hours, then cooling it and grinding it;

pressing the BaCu₃O₄ powder into an arbitrary shape and placing it onto the Y_{1.2}Ba_{0.8}CuO_{3.6} powder;

heating the BaCu₃O₄ and Y_{1,2}Ba_{0,8}CuO_{3,6} in a furnace in flowing O2 at about 950° C. for about 12 hours to form a melt at a shape:powder interface;

holding the furnace at 650° C. for about 4 hours; cooling the furnace to less than 200° C. and removing from the furnace heated materials to bring them to room temperature; and

separating a black melt-produced superconductor 50 from the remaining Y_{1,2}Ba_{0,8}CuO_{3,6} powder in which the superconductor is embedded.

28. The method of claim 27 including the step of removing excess Y_{1.2}Ba_{0.8}CuO_{3.6} powder after separating the black melt-produced supeerconductor for re-use 55 after regrinding.

29. A method for making a superconductor comprising the steps of:

mixing and grinding BaCO3 and CuO with nominal compositions of at least one compound selected 60 from the group consisting of Ba₂Cu₃O₅, BaCu₃O₄, BaCu₄O₅, BaCu₆O₇, and BaCu₁₂O₁₃;

heating the resultant mixture to approximately 850° C. to about 920° C.;

pressing the resultant mixture into a pellet;

placing the pellet on a powder preheated to approximately 900° C. to about 1000° C., comprising RBa₂Cu₃O_{4.5}F₄ wherein R is chosen from the 12

group consisting of: Gd, Ho, Er, Yb, Nd, Sm, Eu, Dy, Tm, and Lu;

heating the pellet and the powder in a furnace to approximately 940° C, to about 1000° C, and

removing a melted interface between the pellet and powder.

30. A method for making a superconductor comprising the steps of:

mixing and grinding BaCO3 and CuO with nominal compositions of at least one compound chosen from the group consisting of Ba₂Cu₃O₅, BaCu₃O₄, BaCu₄O₅, BaCu₆O₇, and BaCu₁₂O₁₃;

heating the resultant mixture to approximately 850° C. to 920° C.;

pressing the resultant mixture into a pellet;

placing the pellet on a powder that has previously been heated to approximately 900° C. to about 1000° C., comprising R_{1.2}Ba_{0.8}CuO_{3.6} wherein R is chosen from the group consisting of: Y, Gd, Er, Ho, Nd, Sm, Eu, Yb, Dy, Tm, and Lu;

heating the pellet and the powder in a furnace to a temperature of greater than or equal to 950° C. and less than or equal to 1000° C. to form a melt at a pellet:powder interface; and

removing a melt-produced superconductor from the powder in which the superconductor is embedded.

31. The method of claim 30 wherein after the pellet and powder are heated to 940°-1000° C. for a predetermined time in the furnace, the temperature of the furnace is lowered to approximately 650° C. and the pellet and mixture remain therein for a predetermined time.

32. The method of claim 30 wherein the pellet and powder are heated in the furnace at a temperature between 940° C. to about 1000° C. for approximately 12 to about 24 hours.

33. A method for making a superconductor comprising the steps of:

mixing and grinding BaCO3 and CuO with nominal compositions of at least one compound chosen from the group consisting of BaCu₃O₅, Ba₂Cu₃O₄, BaCu₄O₅, BaCu₆O₇, and BaCu₁₂O₁₃;

heating the resultant mixture to a temperature of approximately 900° C.;

pressing the resultant mixture into a pellet;

placing the pellet on a powder preheated to approximately 900° C. to about 1000° C., comprising Y₁. 2Ba_{0.8}CuO_{3.6};

heating the pellet and the powder in a furnace to a temperature of greater than or equal to 940° C. and less than or equal to 1000° C. for approximately 12 to about 24 hours to form a melt at a pellet:powder interface;

cooling the furnace to approximately 650° C. and heating the pellet and powder therein;

cooling the furnace to at least approximately 200° C.; and

removing a melted interface between the pellet and powder.

34. A process of making melt-produced superconductor components comprising the steps of:

mixing and grinding BaCO3 and CuO in a nominal composition of BaCu₃O₄;

heating the mixture in air at about 900° C. for about 12 hours, cooling the mixture, and grinding it, and pressing it into a pellet;

mixing and grinding Y2O3, BaCO3, and CuO in a nominal composition of Y_{1.2}Ba_{0.8}CuO_{3.6};

4,857,504 13 heating the mixture in air at about 950° C. for about 12 hours, then cooling it and grinding it; placing the resultant BaCu₃O₄ pellet onto the Y₁. 2Ba_{0.8}CuO_{3.6} powder; heating the BaCu₃O₄ and Y_{1.2}Ba_{0.8}CuO_{3.6} in flowing ⁵ O2 at about 950° C. for about 12 hours to form a melt at the pellet:powder interface; holding the furnace at 650° C. for about 4 hours; cooling the furnace to less than 200° C. and removing 10 from the furnace the heated materials to bring them to room temperature; and separating a black melt-produced superconductor from the remaining Y_{1,2}Ba_{0,8}CuO_{3,6} powder. 35. A process for making melt-produced supercon- 15 ductor components of arbitrary shape comprising the steps of: mixing and grinding BaCO3 and CuO in a nominal composition of BaCu₃O₄; 20 25 30 35 40 45 50

14 heating the mixture in air at about 900° C. for about 12 hours, cooling the mixture, and grinding it; mixing and grinding Y2O3, BaCO3, and CuO in a nominal composition of Y_{1.2}Ba_{0.8}CuO_{3.6} powder; heating the mixture in air at about 950° C. for about 12 hours, then cooling it and grinding it; pressing the BaCu₃O₄ powder into an arbitrary shape and placing it onto the Y_{1.2}Ba_{0.8}CuO_{3.6} powder; heating the BaCu₃O₄ and Y_{1.2}Ba_{0.8}CuO_{3.6} in flowing O₂ at about 950° C. for about 12 hours to form a melt at the pellet:powder interface; holding the furnace at 650° C. for about 4 hours; cooling the furnace to less than 200° C. and removing from the furnace the heated materials to bring them to room temperature; and separating a black melt-produced superconductor from the remaining Y_{1.2}Ba_{0.8}CuO_{3.6} powder in which the superconductor is embedded.

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